Synthesis of 1,3-Selenazol-2(3H)-imines

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Dedicated to Prof. Ludmila L. Rodina, Saint Petersburg, on the occasion of her 70th birthday

The reaction of N, N'-diarylselenoureas **16** with phenacyl bromide in EtOH under reflux, followed by treatment with NH₃, gave N, 3-diaryl-4-phenyl-1, 3-selenazol-2(3H)-imines **13** in high yields (*Scheme 2*). A reaction mechanism *via* formation of the corresponding *Se*-(benzoylmethyl)isoselenoureas **18** and subsequent cyclocondensation is proposed (*Scheme 3*). The N, N'-diarylselenoureas **16** were conveniently prepared by the reaction of aryl isoselenocyanates **15** with 4-substituted anilines. The structures of **13a** and **13c** were established by X-ray crystallography.

Introduction. – Since the discovery of the importance of Se as a microelement in bacteria and animals [1], and the function of the selenoenzyme glutathione peroxidase (GPx) as an antioxidant [2], the interest in organoselenium compounds has increased significantly. As a result, a large series of organoselenium compounds was prepared as GPx mimics, *e.g.*, diorgano di- and triselenides **1** and **2** [3–5], and selenaheterocycles **3–6** [6][7]. In recent years, isoselenocyanates have been shown to be useful building blocks for the synthesis of Se-containing heterocycles [8].



As continuation of our studies toward the use of isoselenocyanates in the synthesis of 1,3-selenazole derivatives [9-14] and with the aim of preparing 1,3-selenazol-

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2(3H)-imines, we investigated the reaction of selenourea derivatives, which are easily accessible from isoselenocyanates, with phenacyl bromide. Whereas the analogous synthesis of 1,3-thiazol-2(3H)-imines 9 via the reaction of thioureas 7 with phenacyl bromides 8, *i.e.*, the *Hantzsch* condensation reaction [15], is well-known [16–21] (*Scheme 1*), no synthesis of the corresponding 1,3-selenazol-2(3H)-imines has been reported so far.



Very recently, *Koketsu et al.* and other groups reported on the synthesis of some 1,3selenazolidine-2-imines [22-25]. For example, *N*,*N'*-diisopropylselenourea (**10**) and 2bromo-2-methylpropanoyl bromide (**11**) in pyridine led to the 2-imino-1,3-selenazolidin-4-one **12**, the structure of which has been determined by X-ray crystallography [22] (*Scheme 1*).

Here, we describe the synthesis of some N,3-diaryl-4-phenyl-1,3-selenazol-2(3*H*)imines **13** in a two step reaction from anilines **14**, aryl isoselenocyanates **15**, and phenacyl bromide.

Results and Discussion. – The selenoureas 16 were prepared conveniently according to the protocol reported in [26]. The reaction of 4-bromophenyl and 4-chlorophenyl isoselenocyanate (15a and 15b, resp.) with 4-substituted anilines 14 in tolouene in the presence of a catalytic amount of Et_3N gave 16a - 16f in high yields as crystalline materials (*Scheme 2*). Then, the solution of equimolar amounts of 16 and phenacyl bromide in EtOH was heated to reflux for 2 h. The cooled mixture was poured into H_2O and treated with aqueous NH_3 to convert the formed hydrobromides 17 to the free 1,3-selenazol-2(3H)-imines 13 (*Scheme 2*). The products were isolated in good yield as colorless solids.

The structures of the products were determined on the basis of the spectroscopic and analytical data, and, in the cases of 13a and 13c, they were established by X-ray crystallography (*Fig.*). In both cases, the five-membered heterocycle is planar, but none of the aryl rings is coplanar with it. In 13c, the 4-(dimethylamino)phenyl residue is attached to the N-atom of the 1,3-thiazole ring, and the 4-bromophenyl moiety is located at the imine N-atom.

According to the NMR spectra, only one product was formed in the reactions with 16a - 16c and 16e, *i.e.*, the formation of 13b, 13c, and 13e occurred regioselectively. On the other hand, the products of the reaction with 16d and 16f consisted of a *ca*. 4:1





Figure. ORTEP Plot [27] of the molecular structure of **13a** and **13c** (50% probability ellipsoids, arbitrary numbering of the atoms)

mixture of two isomeric compounds. For example, the ¹H-NMR spectrum of the product from **16d** and phenacyl bromide showed *singlets* at 6.37 and 6.41 ppm for H-C(5), and 2.27 and 2.32 ppm for Me, each with an intensity ratio of *ca*. 4:1. We assigned the structures **13d** and **13d'** to these products.

Based on the presented results and in analogy to the *Hantzsch* thiazole synthesis, we propose the following reaction mechanism (*Scheme 3*): nucleophilic substitution of Br in phenacyl bromide by the Se-atom of selenourea 16 leads to the isoselenourea



derivative 18, which subsequently undergoes a cyclocondensation to give the hydrobromide of 13 (*cf.* [17]). The plausible intermediate is 19, formed *via* nucleophilic addition of an N-atom of 18 onto the C=O group.

In the case of **18c**, the cyclization occurs selectively *via* the more nucleophilic Natom, bearing the 4-(dimethylamino)phenyl residue. The same selectivity was observed in the cases of the reactions with **16b** and **16e** with a 4-methoxyphenyl substituent. The formation of two isomers of type **13d** and **13d'** in the reaction with **16d** and **16f**, respectively, which bear a 4-methylphenyl substituent in addition to a 4-bromo- or 4chlorophenyl residue, can be explained with the smaller difference of the nucleophilicity of the two N-atoms due to the lower electron-donating properties of the Me group in comparison with the Me₂N and MeO groups.

In conclusion, a convenient synthesis of the little known 1,3-selenazol-2(3H)-imines from diarylselenoureas and phenacyl bromide was developed. This approach can easily be applied for the preparation of differently substituted analogues.

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Experimental Part

1. General. See [28]. M.p.: *Mettler-FP-5* or *Büchi B-450* apparatus; uncorrected. IR Spectra: *Perkin-Elmer-781* or *Perkin-Elmer-1600-FT-IR* spectrophotometer. ¹H- and ¹³C-NMR spectra: *Bruker-AC-300* or *Bruker-ARX-300* instrument (300 and 75.5 MHz, resp.), in (D₆)DMSO or CDCl₃; multiplicity of C-atoms from DEPT spectra. MS: *Finnigan MAT-90* or *Finnigan SSQ-700* instrument (CI (NH₃) or ESI). Elemental analyses were performed by the Mikroanalytisches Laboratorium des Organisch-chemischen Instituts der Universität Zürich.

2. *Starting Materials.* 4-Bromophenyl isoselenocyanate (**15a**) and 4-chlorophenyl isoselenocyanate (**15b**) were prepared according to the protocol described in [29], and *N*,*N*'-diarylselenoureas **16** were prepared according to [26]. All other chemicals were commercially available.

3. Synthesis of N,N'-Selenoureas **16**. 3.1. N,N'-Bis(4-bromophenyl)selenourea (**16a**) [30]. From 1.05 g (4 mmol) of **15a** and 0.69 g (4 mmol) of 4-bromoaniline (**14a**): 1.6 g (92%) of **16a**. M.p. 222.6–224.0° (toluene). Creamy solid. IR: 3175s, 3021s, 2998s, 2832m, 1580m, 1552s, 1524s, 1485s, 1401m, 1333s, 1310s, 1271m, 1254m, 1236m, 1217m, 1171w, 1094w, 1072s, 1016s, 920m, 851m, 827s. ¹H-NMR ((D₆)DMSO): 10.25 (s, 2 NH); 7.52, 7.38 (AA'BB', J_{AB} = 8.8, 8 arom. H). ¹³C-NMR ((D₆)DMSO): 179.1 (s, CSe); 138.9, 117.5 (2s, 4 arom. C); 131.3, 126.7 (2d, 8 arom. CH). CI-MS: 437 (18), 435 (38, [M + 1]⁺), 433 (30), 431 (12), 357 (41), 356 (17), 355 (100), 354 (17), 352 (80). Anal. calc. for C₁₃H₁₀Br₂N₂Se (433.00): C 36.06, H 2.33, N 6.47; found: C 35.52, H 2.31, N 6.39.

3.2. N-(*4-Bromophenyl*)-N'-(*4-methoxyphenyl*)selenourea (**16b**). From 0.5 g (1.9 mmol) of **15a** and 0.23 g (1.9 mmol) of *4-methoxyaniline* (**14b**): 0.57 g (80.3%) of **16b**. M.p. 188.8–190.3° (CH₂Cl₂/hexane). IR: 3175s, 3000s, 2836m, 1608m, 1587w, 1553s, 1527m, 1505s, 1482m, 1463m, 1339s, 1297s, 1247s, 1218m, 1180m, 1166m, 1103m, 1068m, 1031m, 1015m, 921w, 851w, 833s. ¹H-NMR ((D₆)DMSO): 10.08, 9.94 (2s, 2 NH); 7.50, 7.37 (*AA'BB'*, J_{AB} = 8.8, 4 arom. H); 7.26, 6.9I (*AA'BB'*, J_{AB} = 8.9, 4 arom. H); 3.75 (s, MeO). ¹³C-NMR ((D₆)DMSO): 178.8 (s, CSe); 157.0, 139.2, 132.1, 117.2 (4s, 4 arom. C); 131.1, 126.8, 126.6, 113.7 (4d, 8 arom. CH); 55.1 (q, MeO). CI-MS: 387 (3), 385 (4, [*M* + 1]⁺), 383 (2), 307 (41), 306 (17), 305 (100), 304 (13), 303 (60). Anal. calc. for C₁₄H₁₃BrN₂OSe (384.13): C 43.77, H 3.41, N 7.29; found: C 43.83, H 3.50, N 7.34.

3.3. N-(*4-Bromophenyl*)-N'-[*4*-(*dimethylamino*)*phenyl*]*selenourea* (**16c**). From 1.0 g (3.83 mmol) of **15a** and 0.52 g (3.82 mmol) of *4-*(*dimethylamino*)*aniline* (**14c**): 1.3 g (86%) of **16c**. M.p. 171.3 – 174.0° (CH₂Cl₂/hexane). IR: 3172*s*, 3004*s*, 2884*m*, 2842*m*, 2800*m*, 1611*m*, 1575*w*, 1551*s*, 1531*s*, 1515*s*, 1483*m*, 1444*w*, 1397*w*, 1329*s*, 1309*s*, 1247*m*, 1224*m*, 1184*w*, 1165*w*, 1128*w*, 1097*w*, 1064*m*, 1012*m*, 945*m*, 920*m*, 819*s*. ¹H-NMR ((D₆)DMSO): 10.01, 9.74 (2*s*, 2 NH); 7.48, 7.37 (*AA'BB'*, *J_{AB}* = 8.9, 4 arom. H); 7.13, 6.69 (*AA'BB'*, *J_{AB}* = 8.9, 4 arom. H); 2.89 (*s*, 2 Me). ¹³C-NMR ((D₆)DMSO): 178.2 (*s*, CSe); 148.6, 139.4, 127.9, 117.1 (4*s*, 4 arom. C); 131.0, 126.9, 126.1, 112.2 (4*d*, 8 arom. CH); 40.1 (*q*, 2 Me). ESI-MS: 400 (100), 398 (86, $[M + 1]^+$), 396 (48). Anal. calc. for C₁₅H₁₆BrN₃Se (397.17): C 45.36, H 4.06, N 10.58; found: C 45.49, H 4.12, N 10.60.

3.4. N-(*4-Bromophenyl*)-N'-(*4-methylphenyl*)*selenourea* (**16d**). From 0.5 g (1.9 mmol) of **15a** and 0.2 g (1.9 mmol) of *4-methylaniline* (**14d**): 0.6 g (88.2%) of **16d**. M.p. 195.2–197.0 (CH₂Cl₂/hexane). IR: 3167s, 3001s, 2919w, 2832w, 1581m, 1551s, 1530m, 1506m, 1484s, 1398w, 1380w, 1335s, 1238m, 1219m, 1175w, 1106w, 1098w, 1068m, 1014s, 921w, 844w, 822s. ¹H-NMR ((D₆)DMSO): 10.25, 10.10 (2s, 2 NH); 7.58, 7.45 (*AA'BB'*, *J_{AB}* = 8.8, 4 arom. H); 7.33, 7.22 (*AA'BB'*, *J_{AB}* = 8.3, 4 arom. H); 2.36 (*s*, Me). ¹³C-NMR ((D₆)DMSO): 178.6 (*s*, CSe); 139.2, 136.7, 134.6, 117.2 (4s, 4 arom. C); 131.1, 129.0, 126.7, 124.6 (4d, 8 arom. CH); 20.4 (*q*, Me). CI-MS: 371 (4), 369 (6, $[M + 1]^+$), 367 (3), 291 (37), 290 (16), 289 (100), 288 (14), 287 (66). Anal. calc. for C₁₄H₁₃BrN₂Se (368.13): C 45.68, H, 3.56, N 7.61; found: C 45.60, H 3.64, N 7.60.

3.5. N-(4-Chlorophenyl)-N'-(4-methoxyphenyl)selenourea (**16e**). From 0.7 g (3.2 mmol) of **15b** and 0.4 g (3.2 mmol) of **14b**: 1.0 g (90.9%) of **16e**. M.p. 187.9–189.1° (CH₂Cl₂/hexane). IR: 3177s, 3001s, 2836m, 1609m, 1587w, 1554s, 1525m, 1505s, 1486m, 1463m, 1441w, 1401w, 1336s, 1297s, 1248s, 1218m, 1180m, 1166m, 1102m, 1090m, 1031m, 1018m, 922w, 852w, 834s, 821m. ¹H-NMR ((D₆)DMSO): 10.07, 9.94 (2s, 2 NH); 7.42, 7.38 (AA'BB', J_{AB} = 8.9, 4 arom. H); 7.26, 6.91 (AA'BB', J_{AB} = 8.9, 4 arom. H); 3.75 (*s*, Me). ¹³C-NMR ((D₆)DMSO): 178.8 (*s*, CSe); 157.0, 138.7, 132.1, 129.0 (4s, 4 arom. C); 128.1, 126.6, 126.5, 113.7 (4d, 8 arom. CH); 55.1 (*q*, Me). CI-MS: 343 (1), 341 (3, [M + 1]⁺), 339 (1), 263 (14), 262

(11), 261 (73), 260 (18), 259 (100). Anal. calc. for $C_{14}H_{13}ClN_2OSe$ (339.68): C 49.50, H 3.86, N 8.25; found: C 49.68, H 3.97, N 8.32.

3.6. N-(4-Chlorophenyl)-N'-(4-methylphenyl)selenourea (**16f**). From 0.7 g (3.2 mmol) of **15b** and 0.34 g (3.2 mmol) of **14d**: 0.9 g (88.2%) of **16f**. M.p. 195.2–195.9° (CH₂Cl₂/hexane). IR: 3172s, 3002s, 2921m, 2832w, 1584m, 1553s, 1506m, 1489m, 1403w, 1380w, 1332s, 1238m, 1220w, 1175w, 1090s, 1017m, 921w, 849w, 824s. ¹H-NMR ((D₆)DMSO): 10.42, 10.29 (2s, 2 NH); 7.69, 7.62 (AA'BB', $J_{AB} = 8.9$, 4 arom. H); 7.52, 7.41 (AA'BB', $J_{AB} = 8.3$, 4 arom. H); 2.54 (s, Me). ¹³C-NMR ((D₆)DMSO): 178.7 (s, CSe); 138.7, 136.7, 134.6 (3s, 4 arom. C); 129.0, 128.2, 126.4, 124.6 (4d, 8 arom. CH); 20.4 (q, Me). CI-MS: 327 (3), 325 (5, [M + 1]⁺), 323 (3), 247 (14), 246 (12), 245 (74), 244 (19), 243 (100). Anal. calc. for C₁₄H₁₃ClN₂Se (323.68): C 51.95, H 4.05, N 8.65; found: C 51.99, H 4.09, N 8.75.

4. Reaction of N,N'-Diarylselenoureas **16** with Phenacyl Bromide. General Procedure. Approximately 2.5 ml of abs. EtOH were added to a mixture of equimolecular amounts (1 - 2 mmol) of **16** and phenacyl bromide. The mixture was stirred for 2 h under reflux. Then, the soln. was cooled to r.t., poured into cold H₂O, and treated with aq. NH₃. The solid obtained was filtered and dried (MgSO₄). The mother liquor was concentrated *in vacuo*, MeOH was added, and the white solid formed was again filtered and dried.

4.1. *N*₃-*Bis*(4-*bromophenyl*)-4-*phenyl*-1,3-*selenazol*-2(3H)-*imine* (**13a**). From 0.73 g (1.7 mmol) of **16a** and 0.33 g (1.7 mmol) of phenacyl bromide: 0.7 g (77.8%) of **13a**. M.p. 193.1–194.4° (CHCl₃/MeOH). White solid. IR: 3121*w*, 3103*w*, 3084*w*, 3055*w*, 1623*s*, 1574*m*, 1556*m*, 1484*m*, 1475*m*, 1445*w*, 1396*w*, 1349*m*, 1298*m*, 1249*m*, 1144*m*, 1093*w*, 1062*m*, 1013*m*, 1004*m*, 919*m*, 868*m*, 825*m*. ¹H-NMR (CDCl₃): 7.44–7.41 (*m*, 4 arom. H); 7.40–7.22 (*m*, 3 arom. H); 7.21–7.06 (*m*, 4 arom. H); 6.94 (*BB'* of *AA'BB'*, J_{AB} = 8.6, 2 arom. H); 6.43 (*s*, H–C(5)). ¹³C-NMR (CDCl₃): 141.4, 137.5, 132.4, 128.4, 121.8, 117.2 (6*s*, 5 arom. C, C(2), C(4)); 132.8, 132.2, 130.5, 128.61, 128.58, 128.4, 123.2 (7*d*, 13 arom. CH); 98.0 (*d*, C(5)). CI-MS: 539 (7), 538 (12), 537 (52), 536 (27), 535 (100, [*M* + 1]⁺), 534 (32), 533 (83), 532 (25), 531 (35), 530 (10). Anal. calc. for C₂₁H₁₄Br₂N₂Se (533.12): C 47.31, H 2.65, N 5.25; found: C 47.12, H 2.80, N 5.24.

Suitable crystals of 13a for the X-ray crystal structure determination were obtained from Et_2O by slow evaporation of the solvent at 5°.

4.2. N-(4-Bromophenyl)-3-(4-methoxyphenyl)-4-phenyl-1,3-selenazol-2(3H)-imine (13b). From 0.3 g (0.78 mmol) of 16b, 0.15 g (0.78 mmol) of phenacyl bromide, and 15–20 ml of EtOH: 0.35 g (92%) of 13b. M.p. 214.9–215.3° (CH₂Cl₂/MeOH). IR: 3097w, 3055w, 3010w, 2958w, 2928w, 2901w, 2830w, 1614s, 1602s, 1573s, 1557m, 1509s, 1490w, 1479s, 1462w, 1449w, 1434w, 1347m, 1300m, 1285w, 1248s, 1146s, 1096w, 1067m, 1028m, 1006m, 919w, 871m, 830m. ¹H-NMR (CDCl₃): 7.41, 6.96 ($AA',BB', J_{AB} = 8.4$, 4 arom. H); 7.25–7.08 (m, 7 arom. H); 6.80 (BB' of $AA'BB', J_{AB} = 8.7$, 2 arom. H); 6.41 (s, H–C(5)); 3.74 (s, MeO). ¹³C-NMR (CDCl₃): 158.8, 142.1, 131.1, 116.7 (4s, 5 arom. C, C(2), C(4)); 132.6, 129.8, 128.6, 128.3, 128.0, 123.4, 114.3 (7d, 13 arom. CH); 97.5 (d, C(5)); 55.2 (q, MeO). CI-MS: 489 (14), 488 (20), 487 (78), 486 (28), 485 (100, [M + 1]⁺), 484 (26), 483 (46), 482 (15), 481 (14). Anal. calc. for C₂₂H₁₇BrN₂OSe (484.25): C 54.57, H 3.54, N 5.78; found: C 54.15, H 3.59, N 5.75.

4.3. N-(4-Bromophenyl)-3-[4-(dimethylamino)phenyl]-4-phenyl-1,3-selenazol-2(3H)-imine (13c). From 0.6 g (1.5 mmol) of 16c and 0.3 g (1.5 mmol) of phenacyl bromide: 0.55 g (73.3%) of 13c. M.p. 209.3–212.5° (CH₂Cl₂/MeOH). IR: 3106w, 2889w, 2861w, 2817w, 1608s, 1571s, 1557m, 1528s, 1489m, 1475s, 1443m, 1374w, 1338s, 1316w, 1289m, 1248m, 1232m, 1191w, 1138s, 1062s, 1000m, 918w, 860m, 825m, 809m. ¹H-NMR (CDCl₃): 7.40, 6.89 (AA'BB', $J_{AB} = 8.7$, 4 arom. H); 7.26–7.04 (m, 7 arom. H); 6.63 (BB' of AA'BB', $J_{AB} = 8.8$, 2 arom. H); 6.29 (s, H–C(5)); 2.90 (s, 2 Me). ¹³C-NMR (CDCl₃): 143.1, 133.9, 123.1, 116.2 (4s, 5 arom. C, C(2), C(5)); 133.3, 132.6, 129.5, 128.6, 128.5, 128.0, 123.2 (7d, 13 arom. CH); 95.7 (d, C(5)); 40.7 (q, 2 Me). ESI-MS: 500 (46), 498 (100, [M + 1]⁺), 496 (40), 494 (24). Anal. calc. for C₂₃H₂₀BrN₃Se (497.29): C 55.55, H 4.05, N 8.45; found: C 55.73, H 4.08, N 8.38.

Suitable crystals of **13c** for the X-ray crystal structure determination were obtained from Et_2O by slow evaporation of the solvent at 5°.

5.4. N-(4-Bromophenyl)-3-(4-methylphenyl)-4-phenyl-1,3-selenazol-2(3H)-imine (13d) and 3-(4-Bromophenyl)-N-(4-methylphenyl)-4-phenyl-1,3-selenazol-2(3H)-imine (13d'). From 0.3 g (0.81 mmol) of 16d and 0.162 g (0.81 mmol) of phenacyl bromide: 0.3 g (78.9%) of 13d. M.p. 193.4–193.8° (CH₂C1₂/ MeOH). IR: 3103w, 3051w, 3032w, 2918w, 1626s, 1604s, 1574m, 1508m, 1488m, 1478m, 1441w, 1350m, 1292m, 1246m, 1141m, 1065m, 1028w, 1006m, 916m, 868m, 826m. ¹H-NMR (CDCl₃; two isomers 4:1): 7.40, 6.92 (AA'BB', $J_{AB} = 8.6$, 4 arom. H); 7.30–7.05 (m, 9 arom. H); 6.41 (s, H–C(5), minor isomer); 6.37 (s, H–C(5), major isomer); 2.32 (s, Me, minor isomer); 2.27 (s, Me, major isomer). ¹³C-NMR (CDCl₃): 142.0, 137.7, 135.9, 132.9, 116.8 (5s, 5 arom. C, C(2), C(4)); 132.5, 132.0, 130.4, 130.2, 129.6, 128.5, 128.4, 128.2, 128.1, 128.0, 123.2, 121.0 (12d, 13 arom. CH); 96.9 (d, C(5)); 21.2 (s, Me, major isomer); 21.0 (s, Me, minor isomer). CI-MS: 473 (13), 472 (19), 471 (78), 470 (28), 469 (100, [M + 1]⁺), 468 (26), 467 (46), 466 (16), 465 (14). Anal. calc. for C₂₂H₁₇BrN₂Se (468.25): C 56.43, H 3.66, N 5.98; found: C 56.33, H 3.64, N 6.00.

4.5. N-(4-Chlorophenyl)-3-(4-methoxyphenyl)-4-phenyl-I,3-selenazol-2(3H)-imine (**13e**). From 0.6 g (1.77 mmol) of **16e** and 0.35 g (1.76 mmol) of phenacyl bromide: 0.6 g (77.92%) of **13e**. M.p. 200.0–200.7° (CH₂Cl₂/MeOH). IR: 3097w, 3052w, 3013w, 2956w, 2930w, 2903w, 2833w, 1614s, 1603s, 1577s, 1556m, 1509s, 1481s, 1463m, 1451m, 1442m, 1346m, 1309m, 1300m, 1285m, 1247s, 1168m, 1146s, 1087m, 1067m, 1029m, 1008m, 919w, 872m, 832s, 821m. ¹H-NMR (CDCl₃): 7.26, 6.97 (*AA'BB'*, J_{AB} = 8.7, 4 arom. H); 7.19–7.10 (*m*, 7 arom. H); 6.79 (*BB'* of *AA'BB'*, J_{AB} = 8.9, 2 arom. H); 6.35 (*s*, H–C(5)); 3.73 (*s*, Me). ¹³C-NMR (CDCl₃): 158.8, 151.8, 142.0, 133.0, 131.4, 129.0 (6s, 5 arom. C, C(2), C(4)); 130.2, 130.0, 129.7, 128.7, 128.2, 128.1, 122.8, 114.3 (8d, 13 arom. C); 96.5 (*d*, C(5)); 55.3 (*q*, MeO). CI-MS: 445 (6), 444 (11), 443 (44), 442 (26), 441 (100, [*M* + 1]⁺), 440 (20), 439 (49), 438 (18), 437 (17). Anal. calc. for C₂₂H₁₇CIN₂OSe (439.80): C 60.08, H 3.90, N 6.37; found: C 60.01, H 3.95, N 6.38.

4.6. N-(4-Chlorophenyl)-3-(4-methylphenyl)-4-phenyl-1,3-selenazol-2(3H)-imine (13f) and 3-(4-Chlorophenyl)-N-(4-methylphenyl)-4-phenyl-1,3-selenazol-2(3H)-imine (13f'). From 0.5 g (1.5 mmol) of 16f and 0.3 g (1.5 mmol) of phenacyl bromide: 0.6 g (92.3%) of 13f. M.p. 191.1–191.3° (CH₂Cl₂/MeOH). IR: 3108w, 3051w, 3033w, 2918w, 2855w, 1630s, 1620s, 1606s, 1580s, 1509m, 1481s, 1441m, 1350m, 1307w, 1293m, 1246m, 1165w, 1141s, 1105w, 1087m, 1066m, 1028w, 1009m, 916m, 869m, 831m, 821m. ¹H-NMR (CDCl₃, 2 isomers 4 : 1): 7.30–6.95 (m, 13 arom. H); 6.44 (s, H–C(5), minor isomer); 6.39 (s, H–C(5), major isomer); 2.32 (s, Me, minor isomer); 2.27 (s, Me, major isomer). ¹³C-NMR (CDCl₃): 142.1, 142.0, 137.5, 132.8 (4s, 5 arom. C, C(2), C(4)); 130.3, 130.2, 129.7, 129.3, 128.7, 128.6, 128.3, 128.1, 123.0, 121.4 (10d, 13 arom. H); 97.3 (d, C(5)); 21.2 (s, Me, major isomer); 21.0 (s, Me, minor isomer). CI-MS: 429 (6), 428 (11), 427 (44), 426 (26), 425 (100, $[M + 1]^+$), 424 (20), 423 (49), 422 (19), 421 (17). Anal. calc. for C₂₂H₁₇ClN₂Se (423.80): C 62.35, H 4.04, N 6.61; found: C 62.19, H 4.13, N 6.60.

5. X-Ray Crystal-Structure Determination of 13a and 13c (Table, and Figs. 1 and 2)²). All measurements were performed on a Nonius KappaCCD diffractometer [31] using graphite-monochromated MoK_a radiation (λ 0.71073 Å) and an Oxford Cryosystems Cryostream 700 cooler. The data collection and refinement parameters are given in the Table, and views of the molecules are shown in the Figure. Data reduction was performed with HKL Denzo and Scalepack [32]. The intensities were corrected for Lorentz and polarization effects, and absorption corrections based on the multi-scan method [33] were applied. The structure of 13a was solved by direct methods using SIR92 [34], which revealed the positions of all non-H-atoms. The structure of 13c was solved by heavy-atom Patterson methods [35], which revealed the positions of the Br- and Se-atoms. All remaining non-H-atoms were located in a Fourier expansion of the Patterson soln., which was performed by DIRDIF94 [36]. The non-H-atoms were refined anisotropically. All of the H-atoms were placed in geometrically calculated positions and refined using a riding model where each H-atom was assigned a fixed isotropic displacement parameter with a value equal to 1.2 U_{eq} of its parent atom (1.5 U_{eq} for Me groups). The refinement of each structure was carried out on F^2 using full-matrix least-squares procedures, which minimized the function $\Sigma w (F_0^2 - F_c^2)^2$). Corrections for secondary extinction were not applied. In the case of 13c, several large peaks of residual electron density remain unaccounted for and cannot be associated with any logical chemical species. It is presumed that they are artefacts caused by the inferior data quality. Neutral atom scattering factors for non-H-atoms were taken from [37a], and the scattering factors for Hatoms were taken from [38]. Anomalous dispersion effects were included in F_c [39]; the values for f' and f" were those of [37b]. The values of the mass-attenuation coefficients are those of [37c]. All calculations were performed using the SHELXL97 [40] program.

²) CCDC-757788 and -757789 contain the supplementary crystallographic data for this article. These data can be obtained free of charge from the *Cambridge Crystallographic Data Centre via* http://www.ccdc.cam.ac.uk/data_request/cif.

Table.	Crystallog	raphic	Data 1	for (Compounds	13a and	ł 13c
	- /						

	13a	13c	
Crystallized from	Et_2O	Et ₂ O	
Empirical formula	$C_{21}H_{14}Br_2N_2Se$	$C_{23}H_{20}BrN_3Se$	
Formula weight	533.06	497.24	
Crystal color, habit	colorless, needle	colorless, needle	
Crystal dimensions [mm]	$0.05\times0.05\times0.35$	$0.02\times0.05\times0.27$	
Temperature [K]	160(1)	160(1)	
Crystal system	monoclinic	triclinic	
Space group	$P2_{1}/c$	$P\bar{1}$	
Z	4	2	
Reflections for cell determination	45874	64681	
2θ Range for cell determination [°]	4-50	4-55	
Unit cell parameters a [Å]	5.7698(2)	5.9821(2)	
b [Å]	23.1139(8)	10.9232(5)	
c [Å]	14.3302(5)	15.9088(7)	
α [°]	90	100.029(2)	
β [°]	93.394(2)	97.181(3)	
γ [°]	90	99.019(3)	
$V[Å^3]$	1907.8(1)	998.53(7)	
D_x [g cm ⁻¹]	1.856	1.654	
$\mu(MoK_a) [mm^{-1}]$	6.176	3.900	
Scan type	ϕ and ω	ϕ and ω	
$2\theta_{(\max)}$ [°]	50	55	
Transmission factors (min; max)	0.522; 0.744	0.770; 0.928	
Total reflections measured	31960	22568	
Symmetry independent reflections	3356	4556	
Reflections with $I > 2\sigma(I)$	2501	3474	
Reflections used in refinement	3356	4556	
Parameters refined	235	255	
Final $R(F)$ [$I > 2\sigma(I)$ reflections]	0.0360	0.0812	
$wR(F^2)$ (all data)	0.0769	0.2430	
Weighting parameters $[a; b]^a$)	0.0342; 1.1353	0.1645; 1.3205	
Goodness of fit	1.042	1.089	
Final Δ_{\max}/σ	0.001	0.001	
$\Delta \rho (\max; \min) [e Å^{-3}]$	0.44; -0.47	3.62; -0.75	

^a) $w^{-1} = \sigma^2 (F_o^2) + (aP)^2 + bP$ where $P = (F_o^2 + 2F_c^2)/3$.

REFERENCES

- [1] K. Schwarz, C. M. Foltz, J. Am. Chem. Soc. 1957, 79, 3292.
- [2] L. Flohé, G. Loschen, W. A. Günzler, E. Eichele, Hoppe-Seyler's Z. Physiol. Chem. 1972, 353, 987; O. Epp, R. Ladenstein, A. Wendel, Eur. J. Biochem. 1983, 133, 51; A. L. Tappel, Curr. Top. Cell. Regul. 1984, 24, 87; L. Flohé, Curr. Top. Cell Regul. 1985, 27, 473; R. F. Burk, 'Selenium in Biology and Human Health', Springer-Verlag, New York, 1994; K. Aoyama, K. Matsubara, S. Kobayashi, Eur. J. Neurol. 2006, 13, 89; X. G. Lei, W.-H. Cheng, J. P. McClung, Annu. Rev. Nutr. 2007, 27, 41; A. Seiler, M. Schneider, H. Förster, S. Roth, E. K. Wirth, C. Culmsee, N. Plesnila, E. Kremmer, O. Rådmark, W. Wurst, G. W. Bornkamm, U. Schweizer, M. Conrad, Cell Metabolism 2008, 8, 237; B. Kalpakçioğlu, K. Şenel, Clin. Rheumatol. 2008, 27, 141.

- [3] G. Mugesh, A. Panda, H. B. Singh, N. S. Punekar, R. J. Butcher, *Chem. Commun.* 1998, 2227; G. Mugesh, H. B. Singh, *Chem. Soc. Rev.* 2000, 29, 347; G. Mugesh, A. Panda, H. B. Singh, N. S. Punekar, R. J. Butcher, *J. Am. Chem. Soc.* 2001, 123, 839; G. Mugesh, W.-W. du Mont, H. Sies, *Chem. Rev.* 2001, 101, 2125.
- [4] S. R. Wilson, P. A. Zucker, R. R. C. Huang, A. Spector, J. Am. Chem. Soc. 1989, 111, 5936; A. Spector, S. R. Wilson, P. A. Zucker, R. R. Huang, P. G. Raghavan, Lens Eye Tox. Res. 1989, 6, 773; M. Iwaoka, S. Tomoda, J. Am. Chem. Soc. 1994, 116, 2557; M. Iwaoka, S. Tomoda, J. Am. Chem. Soc. 1996, 118, 8077; T. Wirth, Molecules 1998, 3, 164; A. Welter, H. Fisher, L. Christiaens, A. Wendel, E. Etschenberg, Ger. Offen. 1986, Appl. DE 85-3513070; Chem. Abstr. 1987, 106, 084181; A. Spector, S. R. Wilson, P. A. Zucker, U. S. Patent 5,128,365; Chem. Abstr. 1994, 121, P 256039r.
- [5] G. Mugesh, A. Panda, S. Kumar, S. D. Apte, H. B. Singh, R. J. Butcher, Organometallics 2002, 21, 884.
- [6] A. Müller, E. Cadenas, P. Graf, H. Sies, *Biochem. Pharmacol.* 1984, 33, 3235; A. Wendel, M. Fausel, H. Safayhi, G. Tiegs, R. Otter, *Biochem. Pharmacol.* 1984, 33, 3241; M. J. Parnham, S. Kindt, *Biochem. Pharmacol.* 1984, 33, 3247; H. J. Reich, C. P. Jasperse, *J. Am. Chem. Soc.* 1987, 109, 5549; T. G. Back, B. P. Dyck, *J. Am. Chem. Soc.* 1997, 119, 2079.
- [7] T. G. Back, Z. Moussa, J. Am. Chem. Soc. 2003, 125, 13455; T. G. Back, Z. Moussa, J. Am. Chem. Soc. 2002, 124, 12104.
- H. Maeda, N. Kambe, N. Sonoda, S. Fujiwara, T. Shin-ike, *Tetrahedron* 1997, 53, 13667; M. L. Petrov, N. I. Zmitrovich, *Russ. J. Gen. Chem.* 1999, 69, 245; D. R. Garud, M. Koketsu, H. Ishihara, *Molecules* 2007, 12, 504; H. Heimgartner, Y. Zhou, P. K. Atanassov, G. L. Sommen, *Phosphorus, Sulfur Silicon Relat. Elem.* 2008, 183, 840.
- [9] Y. Zhou, A. Linden, H. Heimgartner, Helv. Chim. Acta 2000, 83, 1576.
- [10] P. K. Atanassov, A. Linden, H. Heimgartner, Heterocycles 2003, 61, 569.
- [11] G. L. Sommen, A. Linden, H. Heimgartner, Eur. J. Org. Chem. 2005, 3128.
- [12] G. L. Sommen, A. Linden, H. Heimgartner, Tetrahedron 2006, 62, 3344.
- [13] G. L. Sommen, A. Linden, H. Heimgartner, Helv. Chim. Acta 2007, 90, 1849.
- [14] G. L. Sommen, A. Linden, H. Heimgartner, Helv. Chim. Acta 2008, 91, 209.
- [15] A. Hantzsch, J. H. Weber, Chem. Ber. 1887, 20, 3118; Z. Bartoszewski, Z. Jerzmanovska, Roczn. Chem. 1961, 38, 919; M. J. Korohoda, A. B. Bojarska, Pol. J. Chem. 1984, 58, 447.
- [16] H. Singh, A. S. Ahuja, N. Malhotra, Ind. J. Chem., Sect. B 1980, 19, 1019; V. K. Ahluwalia, B. Mehta, M. Rawat, Ind. J. Chem., Sect. B 1992, 31, 442.
- [17] S. Kasmi, J. Hamelin, H. Benhaoua, Tetrahedron Lett. 1998, 39, 8093.
- [18] G. Kaupp, J. Schmeyers, J. Boy, J. Prakt. Chem. 2000, 342, 269; S. Balalaie, S. Nikoo, S. Haddadi, Synth. Commun. 2008, 38, 2521.
- [19] C. Ciugureanu, M. Ciugureanu, E. D. Murarescu, Rev. Chim. (Bucharest) 2001, 52, 5.
- [20] H. A. Metwally, E. M. Keshk, A. Fekry, H. A. Etman, Phosphorus, Sulfur Silicon Relat. Elem. 2004, 179, 2067.
- [21] K. M. Amin, D. E. Abdel Rahman, Y. A. Al-Eryani, Bioorg. Med. Chem. 2008, 16, 5377.
- [22] M. Koketsu, N. Futoshi, H. Ishihara, Synthesis 2002, 195.
- [23] S. Ueda, H. Terauchi, K. Suzuki, N. Watanabe, Tetrahedron Lett. 2005, 46, 233.
- [24] a) M. Koketsu, T. Kiyokuni, T. Sakai, H. Ando, H. Ishihara, *Chem. Lett.* 2006, 35, 626; b) M. Koketsu, T. Sakai, T. Kiyokuni, D. R. Garud, H. Ando, H. Ishihara, *Heterocycles* 2006, 68, 1607.
- [25] J. Fleischhauer, R. Beckert, W. Günther, S. Kluge, S. Zahn, J. Weston, D. Berg, H. Görls, Synthesis 2007, 2839.
- [26] P. K. Atanassov, A. Linden, H. Heimgartner, Helv. Chim. Acta 2004, 87, 1873.
- [27] C. K. Johnson, 'ORTEP II', Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1976.
- [28] P. K. Atanassov, A. Linden, H. Heimgartner, Heterocycles 2004, 62, 521.
- [29] D. H. R. Barton, S. I. Parekh, M. Tajbakhsh, E. A. Theodorakis, C.-L. Tse, *Tetrahedron* 1994, 50, 639.
- [30] A. Shafiee, F. Assadi, V. I. Cohen, J. Heterocycl. Chem. 1978, 15, 39; V. I. Cohen, Synthesis 1980, 60.
- [31] R. Hooft, KappaCCD Collect Software, Nonius BV, Delft, The Netherlands, 1999.

- [32] Z. Otwinowski, W. Minor, in 'Methods in Enzymology', Vol. 276, 'Macromolecular Crystallography', Part A, Eds. C. W. Carter Jr., R. M. Sweet, Academic Press, New York, 1997, p. 307.
- [33] R. H. Blessing, Acta Crystallogr., Sect. A 1995, 51, 33.
- [34] A. Altomare, G. Cascarano, C. Giacovazzo, A. Guagliardi, M. C. Burla, G. Polidori, M. Camalli, SIR92, J. Appl. Crystallogr. 1994, 27, 435.
- [35] P. T. Beurskens, G. Admiraal, G. Beurskens, W. P. Bosman, S. Garcia-Granda, R. O. Gould, J. M. M. Smits, C. Smykalla, PATTY: The DIRDIF program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands, 1992.
- [36] P. T. Beurskens, G. Admiraal, G. Beurskens, W. P. Bosman, R. de Gelder, R. Israel, J. M. M. Smits, DIRDIF 94: The DIRDIF program system, Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands, 1994.
- [37] a) E. N. Maslen, A. G. Fox, M. A. O'Keefe, in 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 6.1.1.1, p. 477; b) D. C. Creagh, W. J. McAuley, in 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 4.2.6.8, p. 219; c) D. C. Creagh, J. H. Hubbell, in 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 4.2.6.8, p. 219; c) D. C. Creagh, J. H. Hubbell, in 'International Tables for Crystallography', Ed. A. J. C. Wilson, Kluwer Academic Publishers, Dordrecht, 1992, Vol. C, Table 4.2.4.3, p. 200.
- [38] R. F. Stewart, E. R. Davidson, W. T. Simpson, J. Chem. Phys. 1965, 42, 3175.
- [39] J. A. Ibers, W. C. Hamilton, Acta Crystallogr. 1964, 17, 781.
- [40] G. M. Sheldrick, SHELXL97, Program for the Refinement of Crystal Structures, University of Göttingen, Germany, 1997.

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